A SYNTHESIS OF d1-METHYLMYCAMINOSIDE

Seiichi Yasuda and Takeshi Matsumoto

Department of Chemistry, Faculty of Science Hokkaido University, Sapporo, Japan

(Received in Japan 3 September 1969; received in U.K. for publication 22 September 1969)

Mycaminose 1, the common component of magnamycin¹, spiramycin^{1b),2)} and leucomycin³, has been derived from natural sugars by A. C. Richardson⁴⁾ and A. B. Foster et al⁵, as methylgly coside. In this communication we should like to report a simple total synthesis of dl-methylmycaminoside.

Hydroborated compound 3^{6} from 2-ethoxy-6-methyl-3,4-dihydro-2H-pyran 2 was oxidized with H_2O_2 in alkaline solution to give an alcohol 4^{6} in 86 % yield from 2. Reaction of the alcohol 4 with bromine in boiling methanol containing hydrogen chloride for 2 days yielded three bromo compounds, 5_1^{7} m.p. 92 - 92.5° (V_{max}^{1} 3280 cm⁻¹, T_{cDCl}^{2} 8.65 (3H, d, J = 6Hz), 6.45 (3H, s), 5.65 (1H, d, J = 9Hz)), 6, m.p. 44 - 45° (V_{max}^{1} 3340 cm⁻¹, T_{cDCl}^{2} 8.74 (3H, d, J = 6Hz), 6.55 (3H, s), 5.37 (1H, d, J = 3Hz)) and Z, m.p. 76 - 77° (V_{max}^{1} 3240 cm⁻¹, T_{cDCl}^{2} 8.7 (3H, d, J = 6Hz), 6.63 (3H, s), 5.31 (1H, d, J = 1Hz)) in 14, 11 and 47 % yield, respectively. The doublet at T_{cDC}^{2} 5.65 with T_{cDC}^{2} 1 in 5 pointed to the diequatorial disposition of Br and OCH3 in this compound. Since the anomerization of 5 in methanol saturated with hydrogen chloride afforded an equilibrated mixture of 5 and 6 in a ratio of about 1:1, the compound 6 is the anomer.

Treatment of 6 with NaN3 in DMSO at 140° for 24 hours gave unsaturated alcohol 8 in 61% yield. On the other hand, the compound 7, which indicates a doublet at 75.31 with J = 1Hz in nmr spectrum possesses diaxial Br and OCH3, and as expected, readily affords the unsaturated alcohol 8 by treating with NaN3 in DMF at 120 - 125° for 17 hours, in 71% yield. Thus the three bromo compounds are convertible to 8.

4398 No.50

On oxidation with perbenzoic acid in dry benzene at room temperature for 7 days, 8 gave an epoxide 2, m.p. 84 - 84.5° (V mujol 3400, 900 cm⁻¹, C CDCl3 8.72 (3H, d, J = 6Hz), 6.56 (3H, s), 5.15 (1H, d, J = 3Hz)) in 90 % yield. Subsequent treatment of 2 with saturated aqueous solution of dimethylamine at 80° for 9 hours gave dl-methylmycaminoside 10, m.p. 112 -113.5° (V mujol 3540, 3400 cm⁻¹, T CDCl3 8.72 (3H, d, J = 6Hz), 7.5 (6H, S), 7.41 (1H, t, J = 9Hz), 7.01 (1H, t, J = 9Hz), 6.6 (3H, s), 6.25 (1H, q, J = 9 + 4Hz), 5.35 (1H, d, J = 4Hz)) in 75 % yield. The compound 10 was completely identical with the optical active 10^9 , derived from leucomycin, in nmr spectrum.

<u>Acknowledgment</u> The authors are grateful to Dr. Ogura for sending us nmr spectral datum of optical active 10.

REFERENCES AND NOTES

- (a) R. B. Woodward, <u>Angew. Chem.</u>, <u>69</u>, 50 (1957).
 (b) R. B. Woodward,
 L. S. Weiler and P. C. Dutta, <u>J. Am. Chem. Soc.</u>, <u>87</u>, 4662 (1965).
 (c) W. D. Celmer, <u>1bid.</u>, <u>88</u>, 5028 (1966).
- (a) M. E. Kuehne and B. W. Benson, <u>ibid.</u>, <u>87</u>, 4660 (1965).
 (b) C. L. Stevens, G. E. Gutowski, K. G. Taylor and C. P. Bryant, <u>Tetrahedron Letters</u>, 5717 (1966).
 (c) R. Paul and S. Tchelitcheff, <u>Bull. Soc. Chim. France</u>, 734 (1957).
- (a) S. Omura, M. Katagiri, H. Ogura and T. Hata, <u>Chem. Pharm. Bull.</u>
 (<u>Tokyo</u>), <u>16</u>, 1167 (1968). (b) M. Hiramatsu, A. Furusaki, T. Noda,
 K. Naya, Y. Tomiie, I. Nitta, T. Watanabe, T, Take and J. Abe, <u>Bull. Chem. Soc. Japan</u>, <u>40</u>, 2982 (1967). (c) S. Omura, M. Katagiri, T. Hata,
 M. Hiramatsu, T. Kimura and K. Naya, <u>Chem. Pharm. Bull.</u> (<u>Tokyo</u>), <u>16</u>, 1402 (1968).
- 4) (a) A. C. Richardson, <u>Proc. Chem. Soc.</u>, 430 (1961). (b) A. C. Richardson, <u>J. Chem. Soc.</u>, 2758 (1962).

- 5) A. B. Foster, T. D. Inch, J. Lehmann, M. Stacey and J. M. Webber, <u>J. Chem.</u>
 <u>Soc.</u>, 2116 (1962).
- 6) S. Yasuda, T. Ogasawara, S. Kawabata, I. Iwataki and T. Matsumoto, Tetrahedron Letters, in press.
- 7) Satisfactory analyses and spectroscopic data were obtained for all the crystalline new compounds.
- 8) Diequatorial opening of epoxide group has been known; G. Charalambous and E. Percival, <u>J. Chem. Soc.</u>, 2443 (1954).
- 9) T. Hata, S. Omura, M. Katagiri, H. Ogura, T. Ito, K. Naya, T. Noda, J. Abe, T. Watanabe and T. Take, the 10th symposium of the chemistry of natural products, Tokyo, Japan, 1966, Abstracts of Papers. p. 86.